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What is the main distinction between ionic and covalent bonding?

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Objectives e Convert

the mass of a substance to the number of moles of a substance, and the ... A sample of a compound analyzed in a chemistry laboratory consists of 5.34 g of carbon, 0.42 g of hydrogen, and 47.08 g of chlorine. ... Use the circle graphs above to answer the following questions. 1. 2. 3.

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Calculations in
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is the extent to which a
measurement
approaches the true
value of a quantity;
precision is the extent
to which a series of
measurements of the
same quantity made in
the same way agree
with one another. 2.

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CHAPTER 6 REVIEW

Chemical Bonding

SECTION 2 SHORT

ANSWER Answer the following questions in the space provided. 1.

Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the

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MULTIPLE CHOICE 1. C

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2.A 3.D 4.D 5.B 6.D 7.A

8.C 9.A 10.B 11.B 12.C

13.B 14.D 15.B 16.C

17.C 18.B 19.D 20.A

21.B 22.B 23.B 24.A

25.D 26.B 27.C 28.B

SHORT ANSWER 29.

Atoms in a molecular compound share electrons to achieve stability. Atoms in an ionic compound gain or lose

Chemistry Section Review 6.3 - 6

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Review Vocabulary
Chapter 1: Introduction
to Chemistry SECTION
15.1 WATER AND ITS
PROPERTIES (pages
445–449) This section
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properties of water in
the liquid and solid
states and explains
how hydrogen bonding
affects the surface
tension and vapor
pressure of water.

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